

2005 Fall Meeting  
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TI: [Thermodynamic modeling of methane production in Early Archean crust by serpentinization: implications for atmospheric methane.](#)

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AB: Abiotic methanogenesis during the hydration of ultramafic minerals by a CO<sub>2</sub>-bearing fluid may have played an important role in the evolution of Earth's prebiotic atmosphere. If this process was volumetrically significant in the early Earth, it would have provided a robust CH<sub>4</sub> source of atmospheric CH<sub>4</sub> in the Archean. Abiotic methanogenesis would likely have accompanied metamorphic fluid-rock interaction in olivine- and pyroxene-rich rocks in a range of environments, including shallow crustal hydrothermal systems, contact metamorphic systems, and low-grade regional metamorphism. To quantitatively evaluate the conditions necessary for and productivity of abiotic methanogenesis during Archean crustal metamorphism, we are developing a simple, internally consistent thermodynamic model of the fO<sub>2</sub>-buffering ability and speciation generated by ultramafic hydration equilibria in the system Mg-Fe-Si-C-O-H at 300-1300 K and 0.3-20 kbar. In our model, we assume that Fe is rejected by brucite and low temperature hydrous Mg-phyllsilicates. This permits computation of upper limits on fCH<sub>4</sub> and lower bounds on fO<sub>2</sub> for relevant equilibria. Assuming an ideal solid solution in olivine and orthopyroxene, we calculate fO<sub>2</sub> vs. T for a fixed bulk-rock XMg using THERMOCALC (Holland and Powell, 1998) and SUPERFLUID (Belonoshko et al, 1992). As an example, our results show that at P=0.3 kb and olivine XMg =0.9, the serpentinization equilibrium (Mg<sub>0.9</sub>Fe<sub>0.1</sub>)<sub>2</sub>SiO<sub>4</sub> + 1.3 H<sub>2</sub>O + 0.033 O<sub>2</sub> ↔ 0.029 Mg<sub>48</sub>Si<sub>34</sub>O<sub>85</sub>(OH)<sub>62</sub> + 0.388 Mg(OH)<sub>2</sub> + 0.067 Fe<sub>3</sub>O<sub>4</sub> (1) crosses the iron-magnetite buffer at about 375 K (± 10). Also, at ~525K at the same P and XMg, the fO<sub>2</sub> buffered by reaction (1) is two log units above iron-wüstite and four log units below QFM. Using this fO<sub>2</sub>, our speciation calculation for a graphite-saturated C-H-O fluid shows that up to ~700K, CH<sub>4</sub> is the dominant carbon species in a fluid in equilibrium with reaction (1). At ~500 K, 0.3 kb, CH<sub>4</sub>/CO<sub>2</sub> ~ 1010. As with Phanerozoic serpentinization (e.g., Frost, 1984), our modeling demonstrates that interaction of CO<sub>2</sub>-bearing metamorphic fluids with Archean olivine-rich volcanic rocks (e.g., komatiites) has the potential to act as a major CH<sub>4</sub> source. By analogy with CO<sub>2</sub> in the Phanerozoic, the Archean CH<sub>4</sub> cycle may have been strongly influenced by crustal sourcing of CH<sub>4</sub>. Belonoshko, AB, et al. (1992) *Comp. Geosci.* 18, 1267. Frost, BR (1985) *J. Pet.* 26, 31.

Holland, TJB, and Powell, R (1998) J. Metamorphic Geol. 16, 309.  
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